

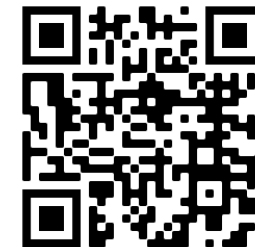
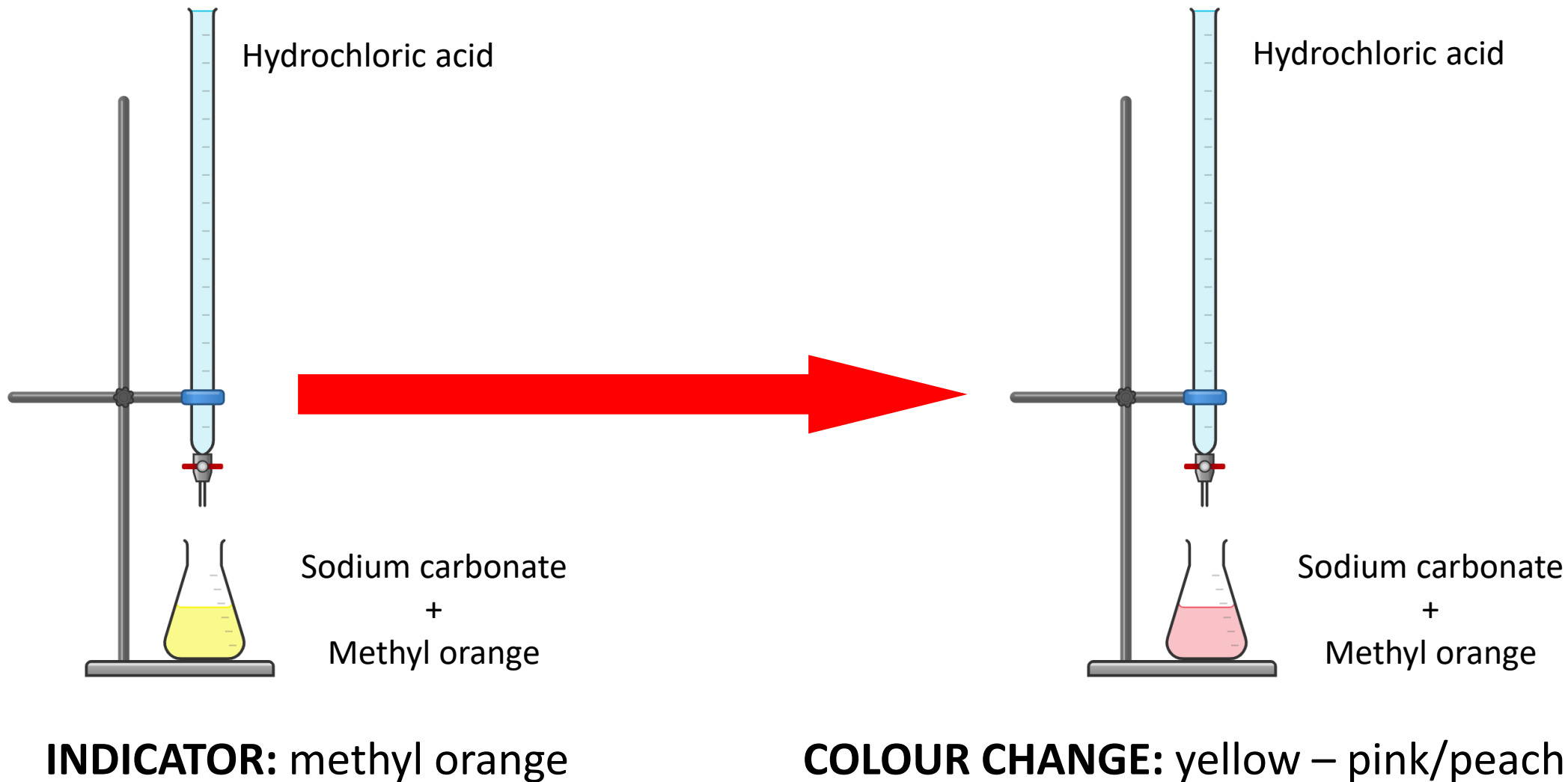


LC Chemistry

# Titration Colour Changes

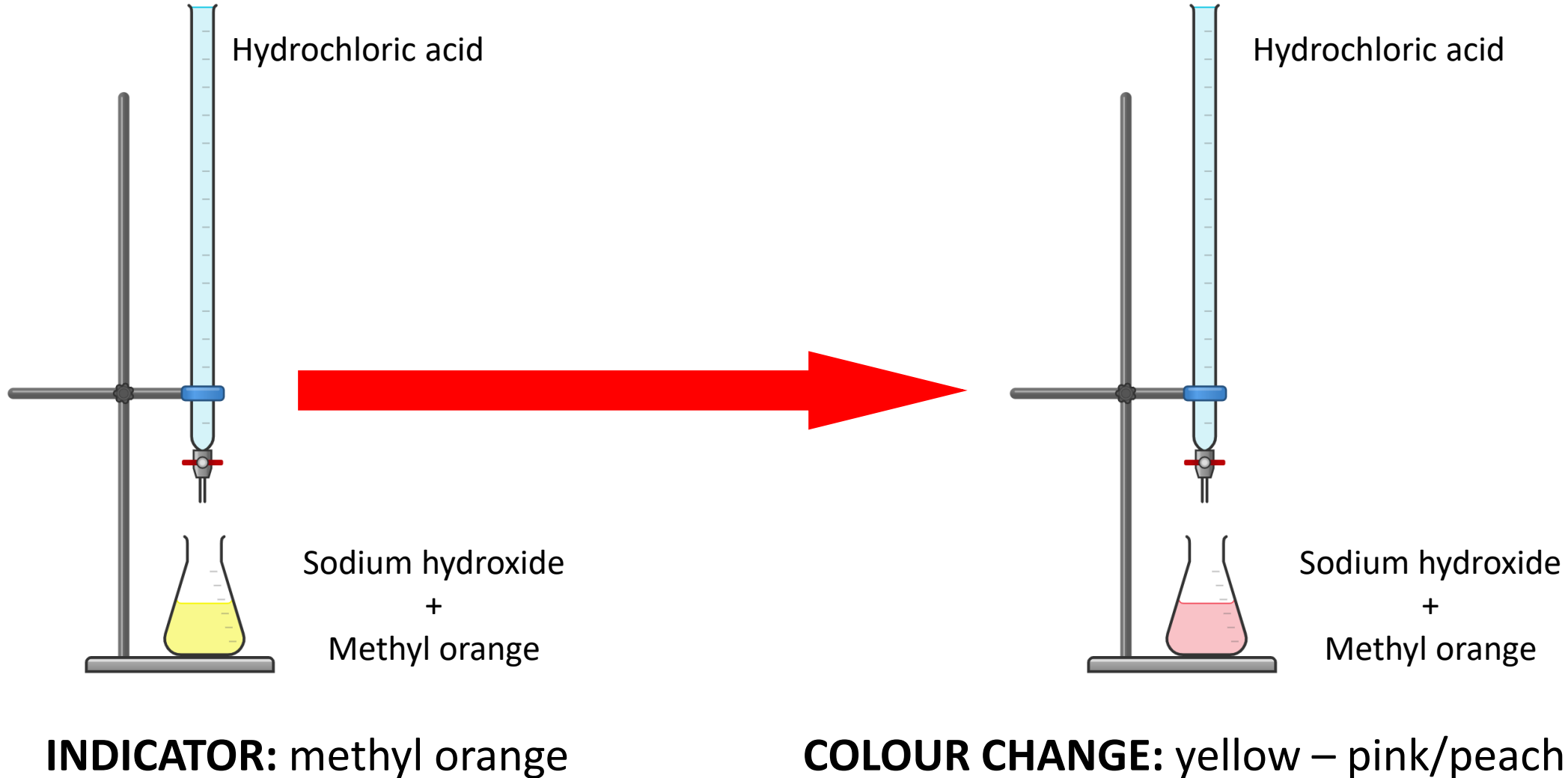
S. Kelleher  
2022

# Standardisation of a hydrochloric acid (HCl) solution using a standard solution of sodium carbonate ( $\text{Na}_2\text{CO}_3$ )



VIDEO LINK

# A hydrochloric (HCl) acid/sodium hydroxide titration (NaOH) to make sodium chloride (NaCl) salt



VIDEO LINK

# Determination of the concentration of ethanoic acid ( $\text{CH}_3\text{COOH}$ ) in vinegar



VIDEO LINK

Vinegar solution

Hydrochloric acid

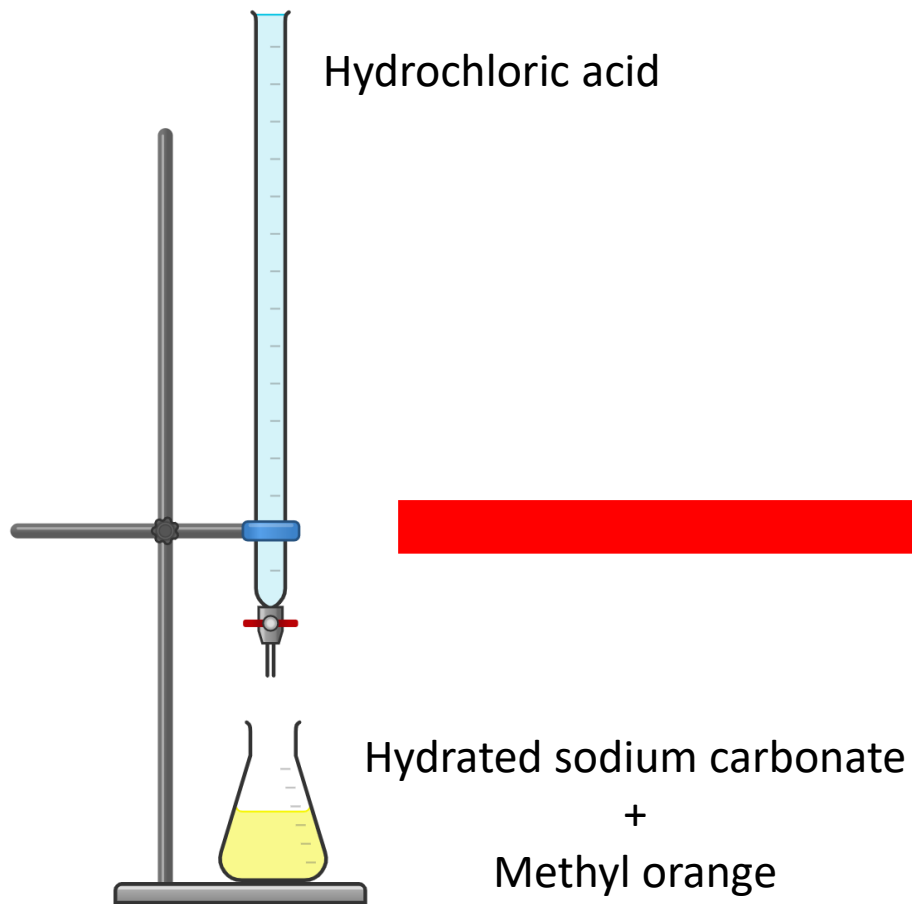
Sodium hydroxide  
+  
Phenolphthalein

Sodium hydroxide  
+  
Phenolphthalein

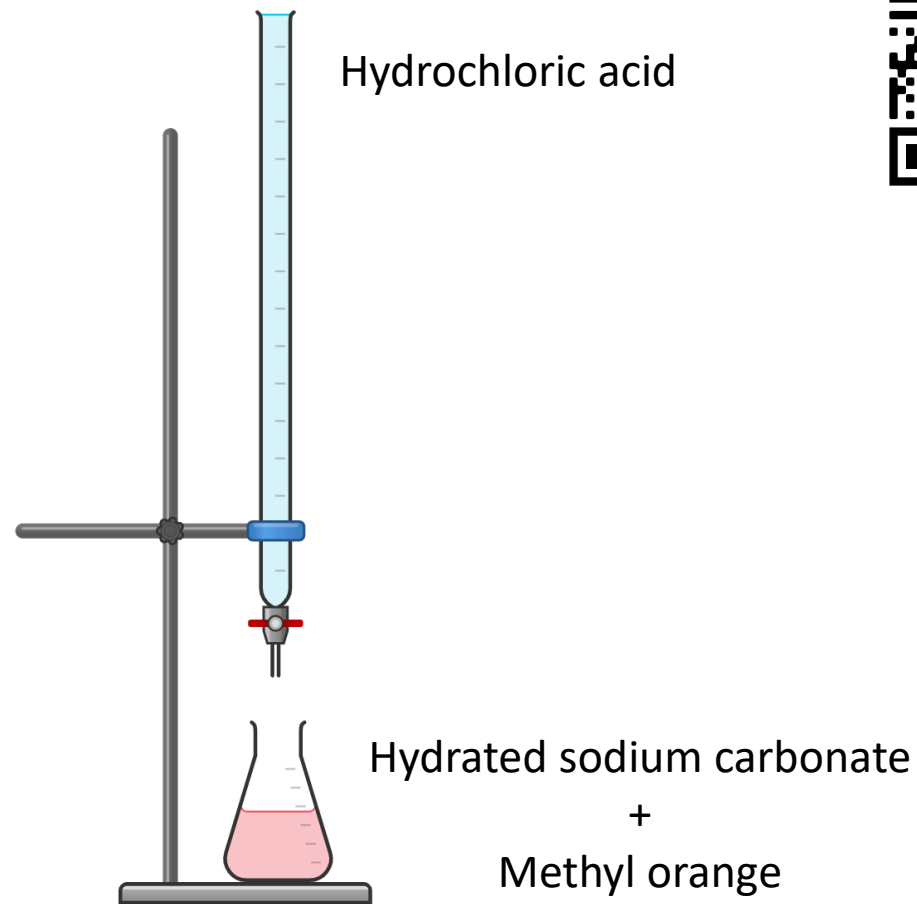
**INDICATOR:** phenolphthalein

**COLOUR CHANGE:** pink - colourless

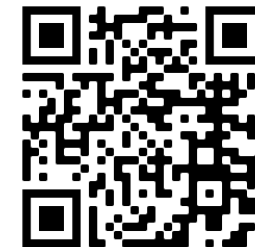
# Determination of water of crystallisation in hydrated sodium carbonate ( $\text{Na}_2\text{CO}_3 \cdot x\text{H}_2\text{O}$ ) – washing crystals



**INDICATOR:** methyl orange



**COLOUR CHANGE:** yellow – pink/peach

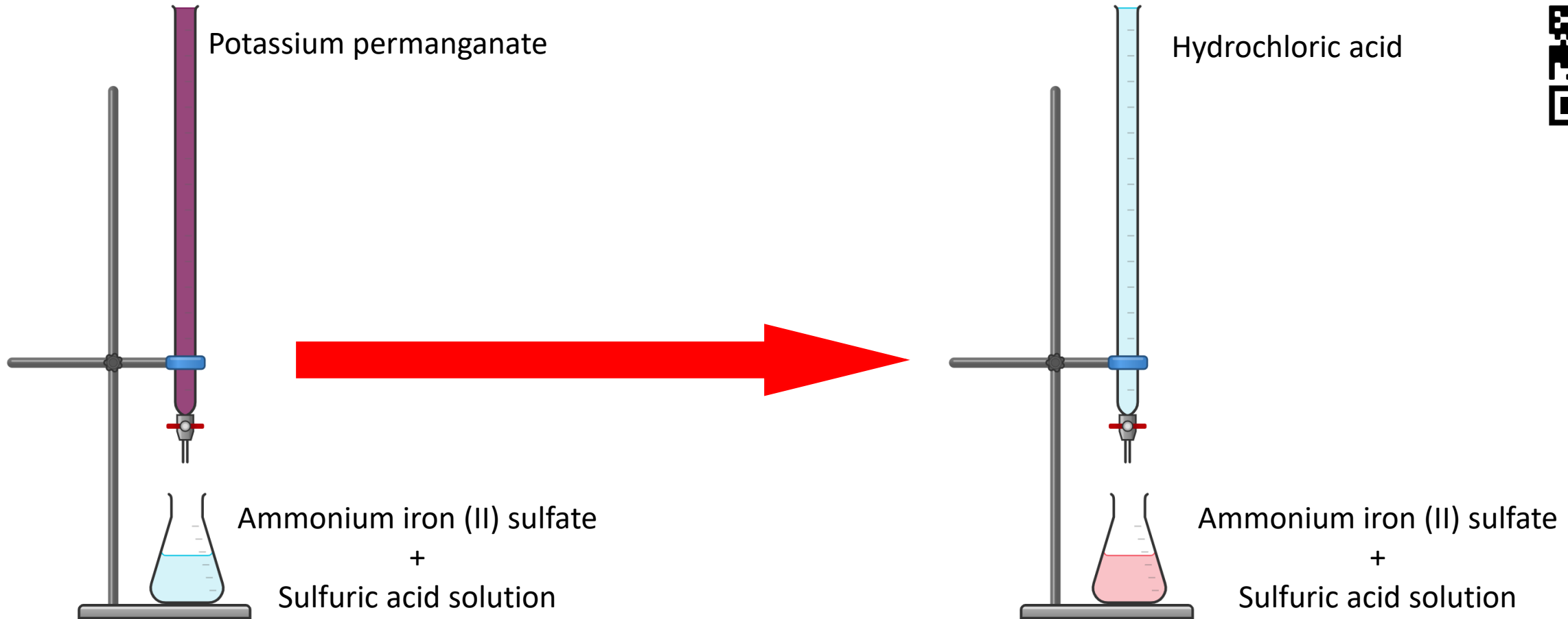


VIDEO LINK

Standardisation of potassium permanganate solution,  $\text{KMnO}_4$ , using a standard solution of ammonium iron (II) sulfate,  $(\text{NH}_4)_2\text{Fe}(\text{SO}_4)_2$



VIDEO LINK

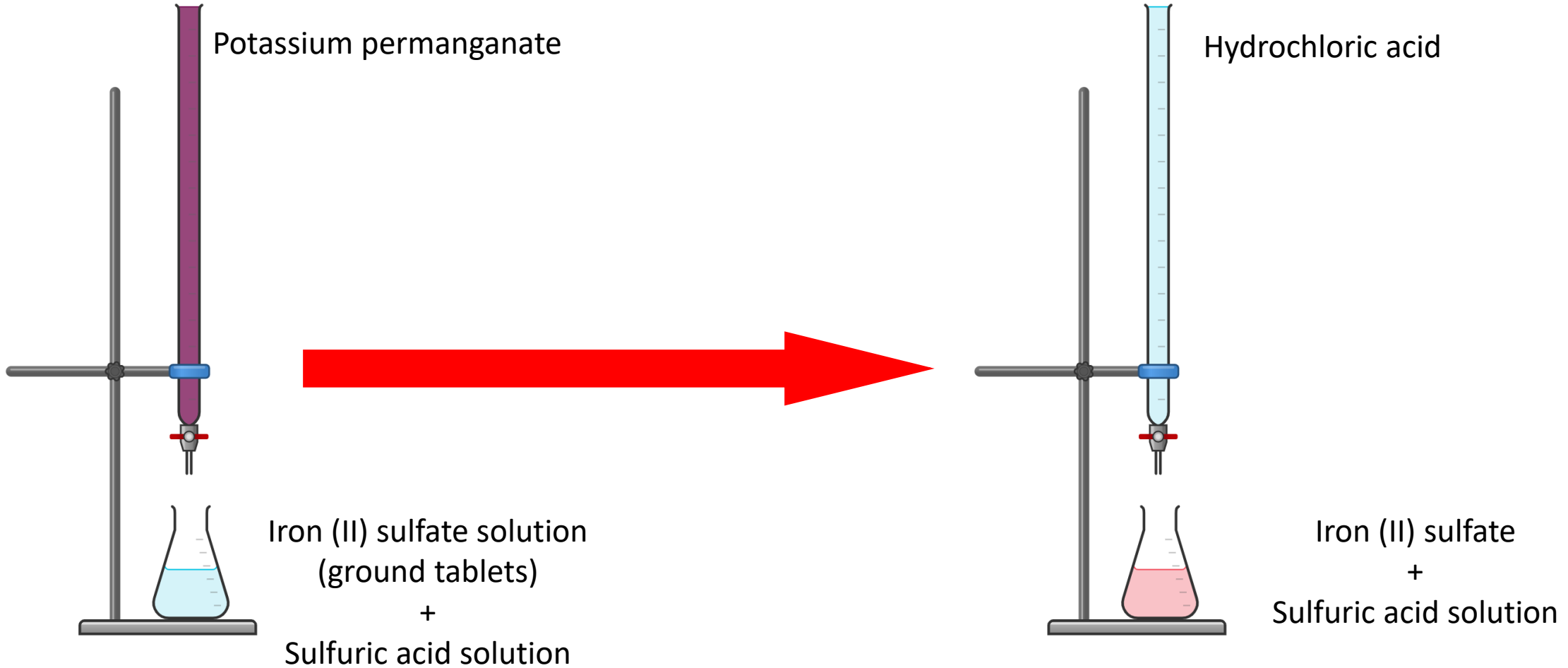


**INDICATOR:** potassium permanganate **COLOUR CHANGE:** colourless – pale pink

# Estimation of iron (II) in an iron tablet using a standard solution of potassium manganate (VII), $\text{KMnO}_4$



VIDEO LINK

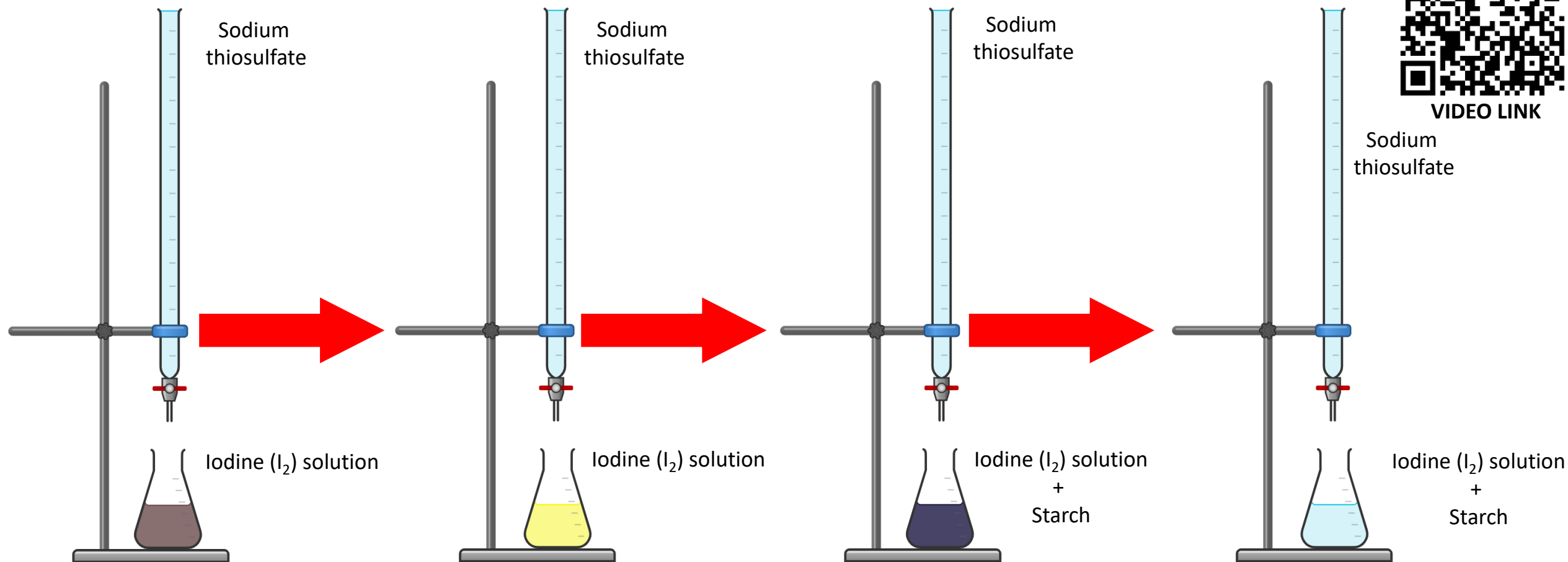


**INDICATOR:** potassium permanganate **COLOUR CHANGE:** colourless – pale pink

# Standardisation of sodium thiosulfate, ( $\text{Na}_2\text{S}_2\text{O}_3$ ), using a standard solution of iodine ( $\text{I}_2$ )



VIDEO LINK



**COLOUR CHANGE 1:** brown to yellow

**INDICATOR:** starch

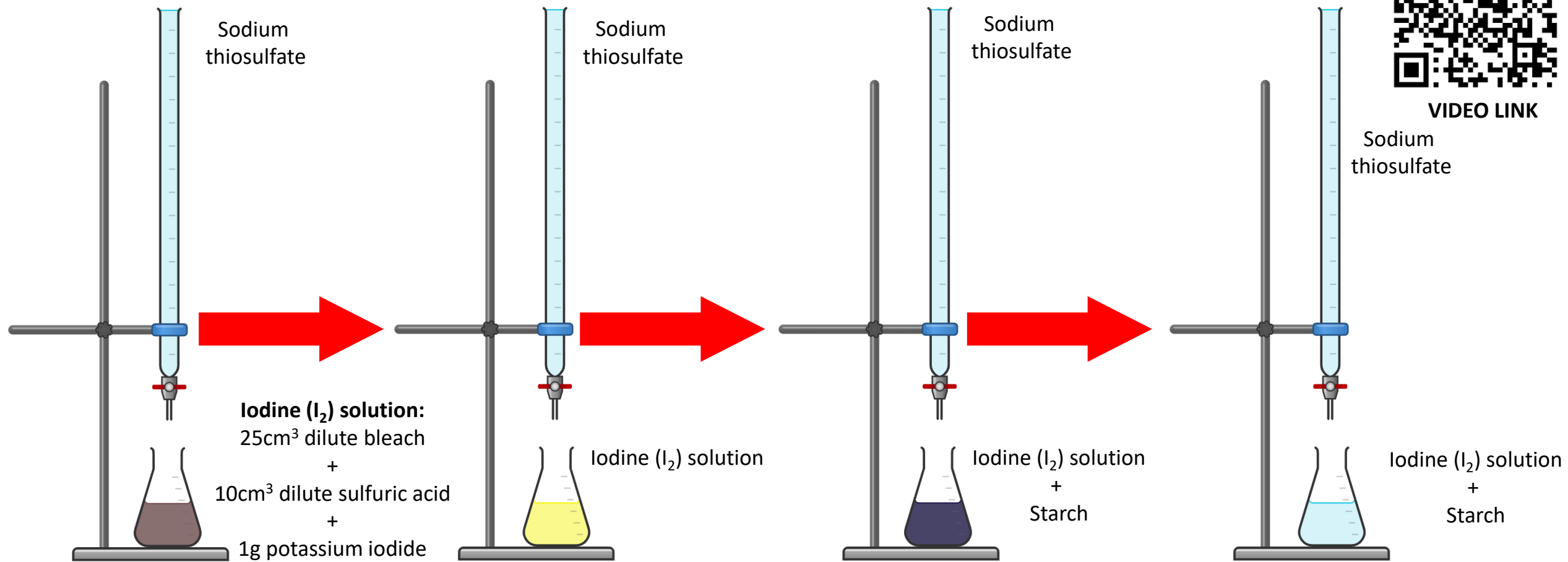
**COLOUR CHANGE 2:** blue-black to colourless

**HIGHER ONLY**

# Determination of the percentage (w/v) of hypochlorite in bleach



VIDEO LINK



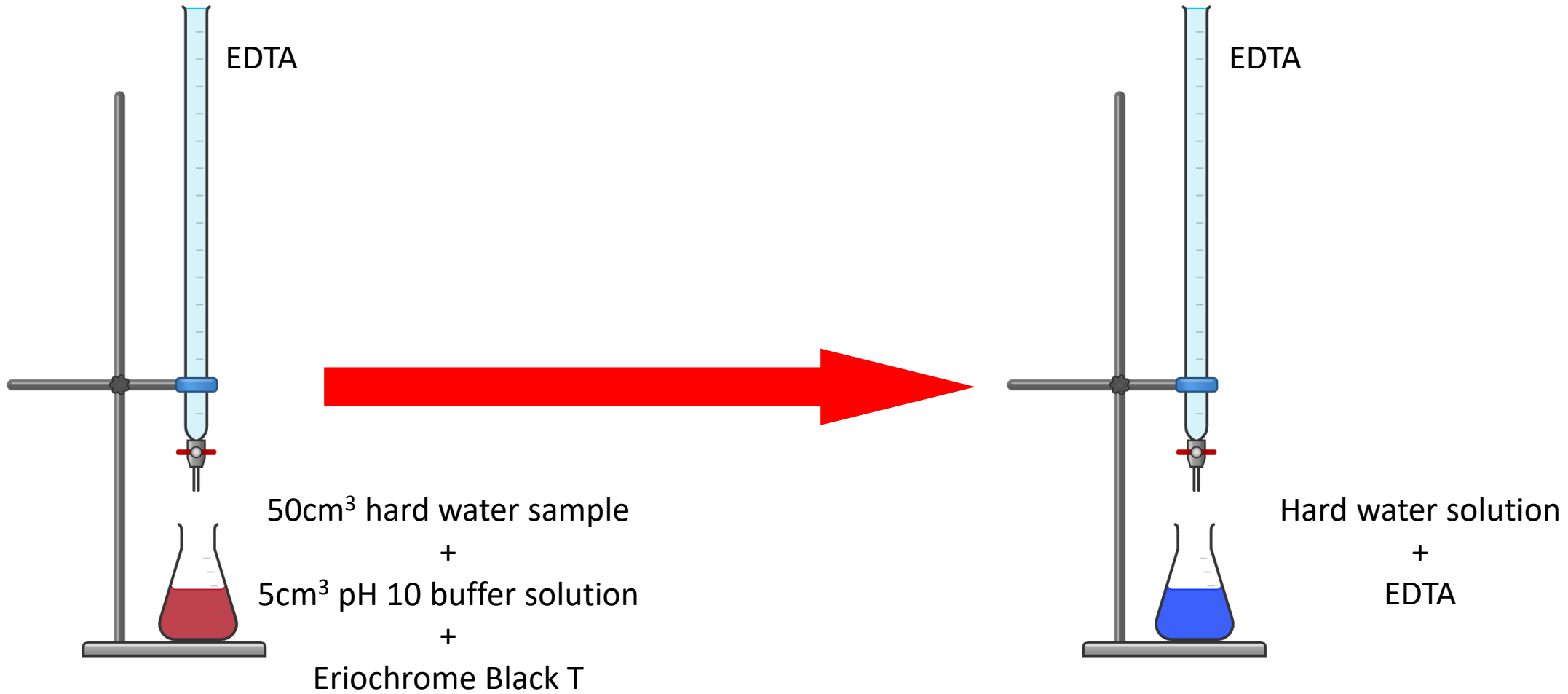
**COLOUR CHANGE 1:** brown to yellow

**INDICATOR:** starch

**COLOUR CHANGE 2:** blue-black to colourless

**HIGHER ONLY**

# Estimation of the total hardness ( $\text{Ca}^{2+}$ ) of a water sample using EDTA



VIDEO LINK

**INDICATOR:** eriochrome black T

**COLOUR CHANGE:** wine red - blue